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		LIO DI ME	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
APPLICATION NO.	FILING DATE		FIRST NAMED IN LINE		2017
10/018,557	04	/11/2002	John Hawkins	MPD316	2017
n.c		01/07/2004		EXAMINER	
7590 01/07/2004 Russell R Stolle				METZMAIER, DANIEL S	
Huntsman Corp					PAPER NUMBER
	Oracion		ART UNIT	PAPER NOMBER	
PO Box 15730				1712	
Austin, TX 78	5761				

DATE MAILED: 01/07/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

No. of the Control of	Office Action Summary	Application No.	Applicant(s)	/ >M \				
	Office Action Summary		HAWKINS, JOHN					
		10/018,557 Examiner	Art Unit					
	Office Accion Cummary	Daniel S. Metzmaier						
	he MAILING DATE of this communication	appears on the cover sh		idress				
Period for F	Reply							
THE MA - Extension after SIX - If the per - If NO per - Failure to	TENED STATUTORY PERIOD FOR RE ILLING DATE OF THIS COMMUNICATIO ns of time may be available under the provisions of 37 CFF (6) MONTHS from the mailing date of this communication iod for reply specified above is less than thirty (30) days, a riod for reply is specified above, the maximum statutory per or reply within the set or extended period for reply will, by state or received by the Office later than three months after the matent term adjustment. See 37 CFR 1.704(b).	IN. R 1.136(a). In no event, however, reply within the statutory minimu riod will apply and will expire SIX	, may a reply be timely filed Im of thirty (30) days will be considered time (6) MONTHS from the mailing date of this come ARANDONED (35 U.S.C. § 133).	ely. communication.				
	esponsive to communication(s) filed on \underline{o}	6 October 2003.						
2a)□ TI	nis action is FINAL . 2b)⊠ T	his action is non-final.						
0,□ 0	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.							
Disposition	ı of Claims							
4)⊠ C	laim(s) <u>1-8</u> is/are pending in the applicati	on.						
48	a) Of the above claim(s) is/are with	ndrawn from considerati	ion.					
5)□ C	claim(s) is/are allowed.							
	claim(s) <u>1-8</u> is/are rejected.							
7) 🗌 C	claim(s) is/are objected to.		ant					
8) C	claim(s) are subject to restriction a	nd/or election requirem	ent.					
Applicatio	n Papers							
9) <u></u> ⊤	he specification is objected to by the Exa	miner.						
10)□ T	10) The drawing(s) filed on is/are: a) accepted or b) objected to by the Examiner.							
Α	Applicant may not request that any objection to	o the drawing(s) be held in	n abeyance. See 37 CFR 1.00(a).	CED 1 121(d)				
F	Replacement drawing sheet(s) including the co	orrection is required if the	drawing(s) is objected to. See 37	DTO-152				
	he oath or declaration is objected to by the	ne Examiner. Note the a	attached Office Action of John	1 10-102.				
Priority ur	nder 35 U.S.C. §§ 119 and 120							
a)☐ * Si 13)☐ Ai sir 37 a)	Acknowledgment is made of a claim for for All b) Some * c) None of: 1. Certified copies of the priority docu 2. Certified copies of the priority docu 3. Copies of the certified copies of the application from the International Bee the attached detailed Office action for cknowledgment is made of a claim for donce a specific reference was included in to CFR 1.78. The translation of the foreign language cknowledgment is made of a claim for dofference was included in the first sentence.	ments have been received ments have been received priority documents have bureau (PCT Rule 17.20) a list of the certified compastic priority under 35 he first sentence of the ge provisional application mestic priority under 35 priority under 3	ved. ved in Application No ve been received in this Nation a)). pies not received. 5 U.S.C. § 119(e) (to a provisio specification or in an Applicati on has been received.	nal application) on Data Sheet. ace a specific				
Attachment		41 🗆	Interview Summary (PTO-413) Paper	No(s)				
2) Notice	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO-9- nation Disclosure Statement(s) (PTO-1449) Paper I	48) 5) 🔲	Notice of Informal Patent Application (PTO-152)				

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1 4 5 or 1 5

DETAILED ACTION

Claims 1-8 are pending.

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 3. Claims 1-8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Albright & Wilson Ltd, EP 0 623 670 A2, in view of van de Pas et al, US 4,530,780, as evidenced by Derwent Abstract, AN 1983-52014k.

Albright & Wilson Ltd (page 6, lines 44 et seq) discloses aqueous based structured surfactant compositions employing a stabilizer system to defloculate the structured surfactant system. Albright & Wilson Ltd (page 7, lines 15-17, and page 8, lines 1-3) disclose the use of monovalent electrolytes with higher polyethoxylates

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employing 7 to 80 moles of ethylene oxide per molecule. Albright & Wilson Ltd (page 8, lines 10-12) teaches alcohol ethoxylates are highly effective stabilizers for systems, which further employ other types of soluble polymers. Albright & Wilson Ltd (page 9, lines 51 et seq, particularly lines 53 and 54-58; page 12, lines 51-53; and page 13, lines 31-38) teaches the use of monovalent electrolytes with C₆₋₂₀ alkyl ethoxylates having preferably 25-75 moles, most preferably 40-55 moles, of ethyleneoxy groups per molecule. Albright & Wilson Ltd (page 11, lines 50 et seq) teaches a number of monovalent electrolytes including alkali metal chlorides and iodides among others. Albright & Wilson Ltd (page 11, lines 11-13) teaches it is often desirable to include potassium salts in the electrolyte for the advantage of reducing viscosities or increasing electrolyte concentrations.

Albright & Wilson Ltd (page 12, lines 39-41) teaches the concentrations of the nonionic stabilizers and page 13, lines 31-32) teach concentrations for the electrolytes. Said concentrations read on the concentrations of claim 3. Albright & Wilson Ltd (throughout) discloses the formation of spherulitic, L₂, and G phase structured surfactants without mention of temperature. Said temperature is concluded to be room temperature, which is below 40°C and 50°C in claims 2 and 3, respectively.

Albright & Wilson Ltd <u>differs</u> from the claims in the exemplified use of a water-soluble thiocyanate compound as the electrolyte.

Albright & Wilson Ltd (page 6, lines 1-27; particularly lines 22-24) discloses EP 0079646 as prior art structured surfactant compositions. The Derwent Abstract An 1983-52014k corresponds to EP 0079646 and shows said document to be a patent

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family member of van de Pas et al, US 4,530,780. The references are <u>combinable</u> because van de Pas et al is cited as prior art structured surfactant compositions in the Albright & Wilson Ltd reference.

van de Pas et al (abstract; page 2, lines 3-40; examples; and claims) discloses the use of alkali metal rhodamides (synonymous with alkali metal thiocyanate) as an auxiliary electrolyte in combination with other electrolytes including polyvalent, wherein said mixture advantageously improves the stability of the structured surfactants. van de Pas et al (examples) discloses nonionic surfactants, which overlap those of the Albright & Wilson Ltd reference.

These references are combinable because they teach structured surfactant compositions and compositions incorporating said surfactant systems. It would have been obvious to one of ordinary skilled in the art at the time of applicants' invention to employ mixtures of electrolytes including as the auxiliary electrolyte, water soluble thiocyanate compound, as taught in the van de Pas et al reference.

Albright & Wilson Ltd (page 17, line 8 to page 21, line 54; and examples) discloses formulating detergent compositions employing the structured surfactant stabilizers as summarized herein above. It would have been obvious to one of ordinary skilled in the art at the time of applicants' invention to employ structured surfactants in the compositions as disclosed in the Albright & Wilson Ltd reference as taught therein. The individual components and the concentrations thereof are broadly taught in the Albright & Wilson Ltd reference.

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Response to Arguments

Applicant's arguments filed October 6, 2003 have been fully considered but they 4. are not persuasive.

Applicant's arguments with respect to claims 1-8 have been considered but are 5. moot in view of the new ground(s) of rejection.

The above new rejection is based on applicants' EP publication in the year of 6. 1994 corresponding to Clapperton, US 6,090,762. The rejection over the Clapperton et al reference has been withdrawn in view of the rejection over the family member document above and deemed to have the same or substantially the same disclosure as the Clapperton et al reference. In view thereof the merits of the Declaration of Roger G M Savage have not been reached at this time.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Daniel S. Metzmaier whose telephone number is (703) 308-0451. The examiner can normally be reached on 9:00 AM to 5:30 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Vasu Jagannathan can be reached on (571) 272-1119. The fax phone number for the organization where this application or proceeding is assigned is (703) 872-9306.

Primary Examiner

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